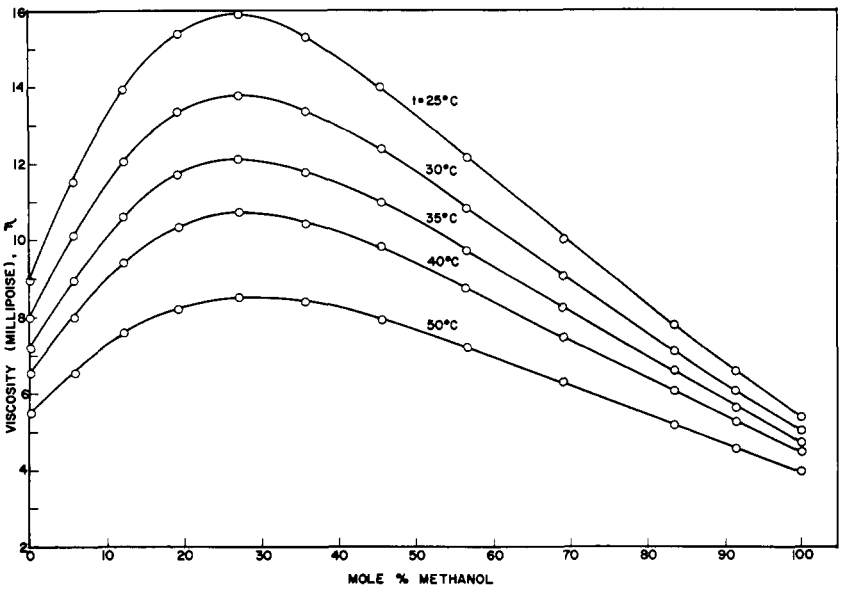

Figure 4. Viscosities of water-methanol mixtures, η, vs. mole per cent methanol

$$\eta \text{ (Millipoises)} = \sum_{i=0}^{4} a_i t^i \qquad (10)$$

This empirical representation by a power series is more precise, within this range of temperature, than that predicted by the theoretically based equations suggested by Gutmann and Simmons (*5*) for viscosity of liquids in general or by Kampmeyer (*10*) and by Innes (*7*) for water and mercury. It is also more precise than that predicted by the equation:

$$\eta = a/(b + t)^n \qquad (11)$$

used in the International Critical Tables (*9*) for methanol. A simple theoretical temperature dependence equation for viscosity of mixtures of liquids has not yet been derived.

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CORRECTION

In "Critical Properties of Mixtures of Normal Paraffin Hydrocarbons" [Doyle O. Etter and W.B. Kay, J. CHEM. ENG. DATA **6**, 409 (1961)], the legends for Figures 2 and 3 should be reversed.

Equation 5 should read:

$$P_c = P_{c_L} + \left[9400 M_L^{-1.71} x_L^{2.06 M_L^{-0.115}} - 557 M_L^{-1.265} \right] (M_{av.} - M_L)$$

Equation 31 should read:

$$\phi_4(P_{cd_4}, x_4) = a_4 \left(\frac{x_4}{1 - x_1 - x_2 - x_3} \right)^{b_4} \left[\frac{M_{av.} - M_1 W_1 - M_2 W_2 - M_3 W_3}{1 - W_1 - W_2 - W_3} - M_4 \right] (1 - x_1 - x_2 - x_3)$$